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## CHRYsin 7-GENTIOBIOSIDE FROM THE FLOWERS OF *SPARTIUM JUNCEUM*

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**Key Word Index**—*Spartium junceum*; Leguminosae; flavonoid glycoside; chrysin 7-gentiobioside.

**Abstract**—Three flavonoids were isolated from the flowers of *Spartium junceum*: the known compounds, chrysin and chrysin 7-glucoside and a new glycoside, chrysin 7-gentiobioside. All three constituents were active in the root growth bioassay.

### INTRODUCTION

*Spartium junceum* L. is known for its diuretic, vasoconstrictor, sedative and other therapeutic properties [1]. There have been several reports [2–4] of B-ring substituted flavonoids and isoflavonoids from this plant. From the flowers of *S. junceum* we now report three unusual flavones (1–3) with no B-ring substituents.

### RESULTS AND DISCUSSION

The flowers of *Spartium junceum* were extracted with acetone and, after evaporation of the solvent, the remaining water was extracted with diethyl ether and then *n*-butanol. Compound 1 was recovered from the ether extract after separation by CC, while 2 and 3 were recovered from the butanol extract.

Compound 1 (mp 288–292°, ethanol) exhibited a molecular ion at  $m/z$  254 (100%) in accord with a flavone structure containing two hydroxyl groups. Its UV spectrum suggested the presence of two free hydroxyl groups at C-5 (bathochromic shifts with aluminium chloride, aluminium chloride–hydrochloric acid and sodium methoxide) and at C-7 (bathochromic shifts with sodium acetate). Compound 1 was identified as chrysin (5,7-dihydroxyflavone) from the spectral data and by comparison with an authentic sample.

Compound 2 exhibited UV maxima in methanol at 269 and 306 nm, bathochromic shifts with aluminium

chloride, aluminium chloride–hydrochloric acid and sodium methoxide and the absence of a shift with sodium acetate, suggested the presence of a free hydroxyl group at C-5. The  $^1\text{H}$  NMR spectrum (DMSO- $d_6$ ) showed, in addition to the aromatic signals resembling those of 1, other signals at  $\delta$  3.2–4.2, suggesting the presence of a sugar substitution on the hydroxyl group at C-7. This hypothesis was confirmed by the  $^{13}\text{C}$  NMR spectrum (DMSO- $d_6$ ) which showed six signals at  $\delta$  60.5, 69.4, 72.9, 76.2, 77.0 and 99.7 for the sugar moiety and by acid hydrolysis which yielded D-glucose and chrysin (1). The  $\beta$ -configuration of glucose was suggested by permethylation of 2. In fact the  $^1\text{H}$  NMR spectrum ( $\text{CDCl}_3$ ) of the permethylated derivative showed a doublet (1H) at  $\delta$  4.95 ( $J = 8$  Hz) attributable to the anomeric proton with a  $\beta$ -configuration [5]. Therefore, 2 was identified as chrysin 7- $\beta$ -D-glucopyranoside.

The UV spectrum of 3 was similar to that of 2, suggesting again the presence of a free hydroxyl group at C-5. The  $^1\text{H}$  NMR and  $^{13}\text{C}$  NMR spectra showed, in the aromatic region, similar signals to those of 2 and more signals in the sugar region suggesting that the sugar moiety linked to the hydroxyl group of chrysin at C-7, was more complex than in 2. Partial acid hydrolysis of 3 yielded 1, 2 and D-glucose, while complete hydrolysis yielded 1 and glucose. The  $\beta$ -configuration of both glycosidic linkages was evident from the presence of two doublets at  $\delta$  5.32 and 5.42 (1H, each,  $J = 8$  Hz) in the  $^1\text{H}$  NMR spectrum (DMSO- $d_6$ ) of permethylated 3. Acid

hydrolysis of permethylated 3 yielded 2,3,4,6-tetramethylglucose and 2,3,4-trimethylglucose, which were identified by TLC comparison with authentic samples. Moreover the hydrogen peroxide oxidation of 3 yielded,  $\beta$ -gentiobiose. From these data, 3 was identified as the new 7-[6-( $\beta$ -D-glucopyranosyl)- $\beta$ -D-glucopyranosyl] derivative of chrysin (chrysin 7-gentiobioside).

Compounds 1–3 were tested in the *Hordeum vulgaris* root growth bioassay [6]. Water solutions (100 ppm) of 1–3 showed an increase of root growth, of 20% for 1; 30% for 2 and 40% for 3. Recently, Nicollier *et al.* [7] observed an activity of chrysin on radish root, whose growth was reduced by 20% at 100 ppm.

## EXPERIMENTAL

**Plant material and general methods.** Flowers of *Spartium junceum* L. were collected near Naples (Pozzuoli) in June. Plant material was verified by comparison with a voucher specimen deposited in the herbarium (NAP) of the Istituto Botanico, Facoltà di Scienze, University of Naples. Mps were uncorr. MS were recorded at 70 eV. Sephadex LH-20 (Pharmacia) and Kieselgel 60 (Merck) were used for chromatography. Precoated Si gel 60 F<sub>254</sub> (Merck) was used for prep. TLC and TLC in: (1) CHCl<sub>3</sub>–MeOH (19:1); (2) Me<sub>2</sub>CO–*n*-BuOH–H<sub>2</sub>O (17:2:1); (3) Me<sub>2</sub>CO–*n*-BuOH–H<sub>2</sub>O (8:1:1); (4) *n*-BuOH–pyridine–H<sub>2</sub>O (8:2:1); (5) Me<sub>2</sub>CO–MeOH–H<sub>2</sub>O (5:4:1); (6) CH<sub>2</sub>Cl<sub>2</sub>–EtOH (19:1); (7) CHCl<sub>3</sub>–MeOH (9:1); and (8) *n*-BuOH–EtOH–H<sub>2</sub>O (5:1:4, upper phase). Compounds were detected in UV light and by spraying with echtronsalz B (Fluka). Sugars and sugar derivatives were detected with aniline phosphate. Commercial *Hordeum vulgaris* seeds were used for bioassay.

**Extraction and isolation.** The flowers (260 g dry wt) of *S. junceum* were extracted with Me<sub>2</sub>CO ( $\times$  3), the extract was concd *in vacuo* and the H<sub>2</sub>O residue extracted, with Et<sub>2</sub>O and then *n*-BuOH, giving on concn a dark brown oil (5.9 g) and an amorphous solid (9.5 g) respectively. The Et<sub>2</sub>O extract, chromatographed on a Si gel column eluted with CHCl<sub>3</sub>–MeOH (49:1), gave 1 (340 mg) with *R<sub>f</sub>* 0.5 on TLC in solvent 1. The *n*-BuOH extract (0.95 g) was triturated with MeOH–H<sub>2</sub>O (4:1), filtered and chromatographed on Sephadex LH-20 (2.5  $\times$  100 cm) column using MeOH–H<sub>2</sub>O (4:1) as eluent. Fractions (10 ml) were collected at 150 ml/hr and investigated by TLC with solvent 2. Compound 3 (70 mg) appeared in fractions 15–22, and 2 (180 mg) in fractions 28–35, with *R<sub>f</sub>*s on TLC in solvent 2 of 0.15 and 0.80 respectively.

**Chrysin 7-glucoside (2).** Mp 213–216° (EtOH aq.). UV  $\lambda_{\max}$  nm: MeOH 269, 306(i); NaOMe 278, 375; AlCl<sub>3</sub> 282, 325, 384; AlCl<sub>3</sub>–HCl 282, 325, 384. <sup>1</sup>H NMR (DMSO-*d*<sub>6</sub>):  $\delta$  3.20–4.20 (ca 7 H, *m*, H-Glc), 6.52 (1H, *d*, *J* = 2 Hz, H-6), 6.95 (1H, *d*, *J* = 2 Hz, H-8), 7.05 (1H, *s*, H-3), 7.70 (3H, *m*, H-3', H-4' and H-5') 8.10 (2H, *m*, H-2' and H-6'). <sup>13</sup>C NMR (DMSO-*d*<sub>6</sub>):  $\delta$  181.8 (*s*, C-4) 163.4 (*s*, C-5), 162.9 (*s*, C-2), 160.8 (*s*, C-7), 156.8 (*s*, C-9), 131.9 (*d*, C-4'), 130.3 (*s*, C-1'), 128.8 (2*d*, C-3' and C-5'), 126.2 (2*d*, C-2' and C-6'), 105.2 (*d* + *s*, C-3 and C-10), 99.8 (*d*, C-1"), 99.5 (*d*, C-6), 94.8 (*d*, C-8), 77.0 (*d*, C-3"), 76.2 (*d*, C-5"), 72.9 (*d*, C-2"), 69.4 (*d*, C-4"), 60.5 (*t*, C-6").

**Chrysin 7-gentiobioside (3).** Mp 185–188° (EtOH). UV  $\lambda_{\max}$  nm: MeOH 270, 306(i); NaOMe 277, 370; AlCl<sub>3</sub> 283, 327, 386; AlCl<sub>3</sub>–HCl 283, 327, 386. <sup>1</sup>H NMR (DMSO-*d*<sub>6</sub>):  $\delta$  3.20–4.20 (ca 14H, *m*, H-Glc), 6.50 (1H, *d*, *J* = 2 Hz, H-6), 6.92 (1H, *d*, *J* = 2 Hz, H-8), 7.05 (1H, *s*, H-3), 7.65 (3H, *m*, H-3', H-4' and H-5'), 8.06 (2H, *m*, H-2' and H-6'). <sup>13</sup>C NMR (DMSO-*d*<sub>6</sub>):  $\delta$  182.0 (*s*, C-

4), 169.7 (*s*, C-5), 162.9 (*s*, C-2), 162.7 (*s*, C-7), 157.3 (*s*, C-9), 131.5 (*d*, C-4'), 130.4 (*s*, C-1'), 128.7 (2*d*, C-3' and C-5'), 125.9 (2*d*, C-2' and C-6'), 115.9 (*d*, C-8), 105.4 (*d* + *s*, C-3 and C-10), 100.0 (*d*), 99.8 (*d*), 99.6 (*d*, C-6), 76.9 (*d*), 76.3 (*d*), 76.1 (*d*), 73.7 (*d*), 72.9 (2*d*), 69.5 (2*d*), 63.1 (*t*, C-6"), 60.5 (*t*, C-6").

**Hydrolysis of 2 and 3.** Compounds 2 (15 mg) and 3 (10 mg) were refluxed with 2 N HCl (3 ml) for 2 hr. The reaction mixtures were evaporated *in vacuo* and analysed on TLC with solvents 1, 3 and 4. Both mixtures gave 1 and glucose.

**Partial hydrolysis of 3.** Compound 3 (10 mg) was refluxed with 1 N HCl (3 ml) for 30 min; the concd mixture was analysed on TLC with solvents 1–4 giving 1, 2 and glucose.

**H<sub>2</sub>O<sub>2</sub> oxidation of 3.** A soln of 3 (10 mg) in 36% H<sub>2</sub>O<sub>2</sub> (10 ml) was kept at room temp. for 1 week. In order to decompose the excess of H<sub>2</sub>O<sub>2</sub>, 5% Pd–C was added; after filtration of catalyst and concn *in vacuo* of filtrate,  $\beta$ -gentiobiose was recovered and identified on TLC with solvents 3 and 5, by comparison with an authentic sample.

**Permethylation.** Permethylation of 2 (30 mg) and 3 (30 mg) was carried out with MeI and Ag<sub>2</sub>O in DMF. The products were purified on prep. TLC with solvent 1 to give permethylated 2 (24 mg) and permethylated 3 (26 mg).

**Permethylated 2.** Mp 187–190° (MeOH). <sup>1</sup>H NMR (CDCl<sub>3</sub>):  $\delta$  3.40 (3H, *s*, OMe), 3.57 (3H, *s*, OMe), 3.67 (6H, *s*, OMe), 3.96 (3H, *s*, OMe), 4.95 (1H, *d*, *J* = 8 Hz, H-1"), 6.53 (1H, *d*, *J* = 2 Hz), 6.68 (1H, *s*), 6.73 (1H, *d*, *J* = 2 Hz), 7.50 (3H, *m*), 7.86 (2H, *m*).

**Permethylated 3.** Mp 173–178° (MeOH). <sup>1</sup>H NMR (DMSO-*d*<sub>6</sub>):  $\delta$  3.38 (9H, *s*, OMe), 3.46 (3H, *s*, OMe), 3.56 (3H, *s*, OMe), 3.57 (6H, *s*, OMe), 3.86 (3H, *s*, OMe), 5.32 (1H, *d*, *J* = 8 Hz, H-1"), 5.42 (1H, *d*, *J* = 8 Hz, H-1"), 6.59 (1H, *d*, *J* = 2 Hz), 6.80 (1H, *s*), 6.94 (1H, *d*, *J* = 2 Hz), 7.59 (3H, *m*), 8.04 (2H, *m*).

**Hydrolysis of permethylated 2 and permethylated 3.** The hydrolyses were performed as for 2 and 3. The sugar derivatives were identified on TLC in solvents 6–8 by comparison with authentic samples.

**Bioassay.** The root growth bioassay technique was essentially that of ref. [6] in which samples were dissolved in 0.5 ml EtOH or MeOH and diluted with H<sub>2</sub>O at 100 ppm. The soln was poured into 10 Petri dishes, with Whatman grade 113 paper as support and 10 *H. vulgaris* seeds placed on each plate. After 4 days at 20° in the dark, the length of the main root was measured and compared with that of control plants.

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